

Roberto A. Steiner,<sup>a\*</sup>  
Ursula Frerichs-Deeken<sup>b</sup> and  
Susanne Fetzner<sup>b</sup>

<sup>a</sup>Randall Division of Cell and Molecular  
Biophysics, King's College London, New Hunt's  
House, Guy's Campus, London SE1 1UL,  
England, and <sup>b</sup>Institut für Molekulare  
Mikrobiologie und Biotechnologie,  
Westfälische Wilhelms-Universität Münster,  
D-48149 Münster, Germany

Correspondence e-mail:  
roberto.steiner@kcl.ac.uk

Received 9 January 2007  
Accepted 21 March 2007

## Crystallization and preliminary X-ray analysis of 1*H*-3-hydroxy-4-oxoquinoline 2,4-dioxygenase from *Arthrobacter nitroguajacolicus* R61a: a cofactor-devoid dioxygenase of the $\alpha/\beta$ -hydrolase-fold superfamily

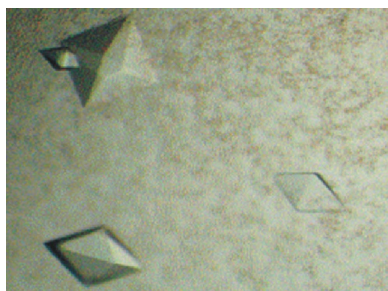
1*H*-3-Hydroxy-4-oxoquinoline 2,4-dioxygenase (HOD) is a cofactor-devoid dioxygenase that is involved in the anthranilate pathway of quinoline degradation. HOD has been proposed to belong to the  $\alpha/\beta$ -hydrolase-fold superfamily of enzymes. N-terminally His<sub>6</sub>-tagged HOD has been crystallized by the hanging-drop vapour-diffusion method using sodium/potassium tartrate as a precipitant and CuCl<sub>2</sub> as an additive. The structure was solved by the single anomalous dispersion (SAD) technique using data collected to 3.5 Å resolution at the Cu absorption peak wavelength. The crystals belong to the primitive tetragonal space group *P*4<sub>3</sub>2<sub>1</sub>2, with unit-cell parameters *a* = *b* = 153.788, *c* = 120.872 Å.

### 1. Introduction

In several aerobic metabolic pathways, O<sub>2</sub> is incorporated into organic compounds through monooxygenase- or dioxygenase-catalyzed reactions (Hayaishi, 1974). For instance, oxygenation is often used to make lipids and particularly aromatic molecules amenable to further biochemical transformations. 1*H*-3-Hydroxy-4-oxoquinoline 2,4-dioxygenase (HOD; EC 1.13.11.48) from *Arthrobacter nitroguajacolicus* R61a is a 32 kDa enzyme that is involved in the anthranilate pathway of quinoline degradation (Bauer *et al.*, 1996). It catalyses the O<sub>2</sub>-dependent N-heteroaromatic ring cleavage of 1*H*-3-hydroxy-4-oxoquinoline to *N*-acetylanthranilate and carbon monoxide (Fig. 1).

Sequence analysis and secondary-structure prediction indicate that HOD belongs to the large superfamily of  $\alpha/\beta$ -hydrolase-fold enzymes (Fischer *et al.*, 1999; Nardini & Dijkstra, 1999). HOD displays no sequence similarities to other known oxygenases, except for the 1*H*-3-hydroxy-4-oxoquinoline 2,4-dioxygenase (QDO; EC 1.13.11.47) involved in 1*H*-4-oxoquinoline degradation by *Pseudomonas putida* 33/1 (Qi *et al.*, 2007). QDO catalyses a reaction very similar to that catalysed by HOD; namely, the dioxygenolytic N-heteroaromatic ring cleavage of 1*H*-3-hydroxy-4-oxoquinoline to *N*-formylanthranilate and carbon monoxide (Bauer *et al.*, 1996). HOD and QDO share 37% identity at the sequence level and these two enzymes are the only dioxygenases that have been proposed to be members of the  $\alpha/\beta$ -hydrolase-fold enzymes.

Oxygenases usually depend on a transition metal and/or organic cofactors for activity. This is a consequence of the fact that a direct reaction between the triplet ground-state O<sub>2</sub> and singlet ground-state substrates to produce singlet-state products implies a violation of the conservation of the total angular momentum (Hamilton, 1974). It is therefore a low-probability event. The presence of cofactors overcomes the spin-forbiddenness of the process. The mechanism used by nonhaem iron-dependent (Que, 1999), copper-dependent (Steiner *et al.*, 2002), flavin-dependent and pterin-dependent (Massey, 1994; Palfey *et al.*, 1995) oxygenases have been studied and various reviews are available that discuss the general mechanisms of oxygen and substrate activation in enzymatic oxygenation reactions. Biochemical and spectroscopic studies have shown that neither HOD nor QDO



© 2007 International Union of Crystallography  
All rights reserved

contain organic cofactors or stoichiometric amounts of any metal (Bauer *et al.*, 1996; Fetzner, 2002). The mechanism(s) employed by these dioxygenases are therefore very interesting from the viewpoint of fundamental enzymology.

The availability of three-dimensional information should increase our understanding of the catalytic mechanism employed by these intriguing cofactor-devoid dioxygenases as well as provide direct evidence about their fold. In this account, we report the crystallization and preliminary X-ray analysis of HOD.

## 2. Materials and methods

### 2.1. Cloning and expression

The *hod* gene from genomic DNA of *A. nitroguajacolicus* Rü61a was isolated and cloned by PCR amplification as described in Frerichs-Deeken *et al.* (2004). For protein overexpression in *Escherichia coli*, the target gene was inserted into the pQE30 vector (Qiagen) at the *Bam*HI and *Sal*I restriction sites. An N-terminal His<sub>6</sub> fusion tag of sequence MRGSHHHHHHGS was added to the gene product in order to facilitate protein purification. The recombinant vector containing the target gene was transformed in competent *E. coli* M15[pREP4] cells by electroporation.

Transformed cells were grown overnight in 10 ml Luria–Bertani (LB) medium supplemented with 50 µg ml<sup>-1</sup> kanamycin and 100 µg ml<sup>-1</sup> ampicillin at 298 K. 400 µl of the overnight starter culture was then transferred into 400 ml LB medium supplemented with antibiotics and incubated at 310 K until the OD at 600 nm reached 0.5. After decreasing the temperature to 301 K, expression of His<sub>6</sub>-HOD was induced with 0.5 mM IPTG; the cells were allowed to grow for a further 20 h and were then harvested by centrifugation at 3500g and 298 K for 15 min.

### 2.2. Purification

Cells were resuspended in 50 mM Tris–HCl pH 8.0, 200 mM NaCl, 20 mM imidazole supplemented with 1× protease-inhibitor cocktail (Calbiochem) lysis buffer and lysed by pulsed sonication on ice. Cell debris was pelleted by centrifugation at 44 000g and 277 K for 45 min. The supernatant was filtered through 0.22 µm filters and loaded onto a 5 ml Ni-loaded His-Trap (GE Healthcare) column previously equilibrated with 50 mM Tris–HCl pH 8.0, 200 mM NaCl, 20 mM imidazole and washed with this buffer until a constant A<sub>280</sub> was observed. Protein was eluted with a gradient to 50 mM Tris–HCl pH 8.0, 200 mM NaCl, 500 mM imidazole and fractions containing HOD were pooled.

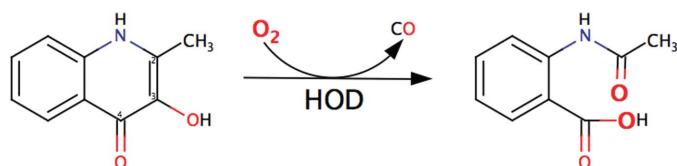
After dialysis overnight against 50 mM Tris–HCl pH 8.0, 5 mM KCl, 2 mM EDTA, the His<sub>6</sub>-HOD solution was loaded onto a Resource-Q (GE Healthcare) ion-exchange column previously equilibrated with dialysis buffer. HOD was eluted with a gradient to 50 mM Tris–HCl pH 8.0, 600 mM KCl, 2 mM EDTA. The protein was concentrated and further equilibrated in 20 mM Tris–HCl pH 7.5,

100 mM NaCl, 2 mM EDTA, 1 mM DTT (storage buffer) for storage at 193 K.

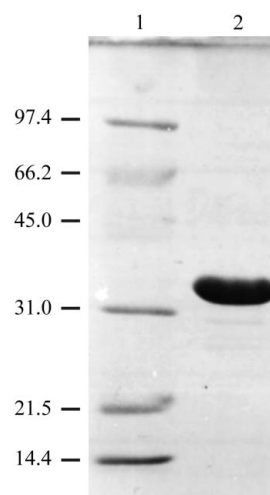
Frerichs-Deeken *et al.* (2004) have shown that HOD has a tendency to form a mixture of monomeric and dimeric species and that dimerization can be prevented by mutating Cys69 to Ser. The HOD C69S variant has catalytic properties that are identical to those of wild-type HOD (Frerichs-Deeken *et al.*, 2004). Therefore, we produced the His<sub>6</sub>-HOD C69S variant as well as the inactive His<sub>6</sub>-HOD C69S/H251A double variant using the QuikChange (Stratagene) mutagenesis system. The His<sub>6</sub>-HOD C69S/H251A double variant is catalytically inactive owing to the His→Ala substitution of the essential residue His251 (Frerichs-Deeken *et al.*, 2004). Both His<sub>6</sub>-HOD variants were purified according to the protocol described above. An SDS–PAGE gel of purified His<sub>6</sub>-HOD C69S/H251A is shown in Fig. 2. On average, the typical yield of pure protein material is approximately 30 mg per litre of culture.

### 2.3. Crystallization

Owing to their more homogeneous behaviour in solution, crystallization trials were mainly carried out using the His<sub>6</sub>-HOD variants described in §2.2. Crystallization conditions were established with the hanging-drop vapour-diffusion technique using 24-well Linbro plates. All experiments were carried out at a constant temperature of 291.15 K using a protein:precipitant ratio of 1.0. Initial crystallization trials of His<sub>6</sub>-HOD C69S/H251A at concentrations of up to 50 mg ml<sup>-1</sup> in storage buffer using various commercial screens (Crystal Screen, Crystal Screen 2, Index, SaltRx, PEG/Ion from Hampton Research) as well as noncommercial versions of the Clear Crystal Strategy 1 and Clear Crystal Strategy 2 screens (Brzozowski & Walton, 2001) failed to produce crystals. However, analysis of the results obtained from the crystallization trials using the various kits indicated salts as being more likely to produce crystals compared with organic precipitants. Crystallization conditions containing salts generally gave light precipitates, whereas organic precipitants tended to yield amorphous precipitates. Amorphous precipitates were also often observed in crystallization conditions buffered at pH values below 5.0–5.5. Screening of several salt precipitants (ammonium citrate, ammonium tartrate, sodium/potassium tartrate, sodium



**Figure 1**  
2,4-Dioxygenolytic ring cleavage of 1H-3-hydroxy-4-oxoquinoline catalysed by HOD. Oxygen atoms from dioxygen are shown in larger bold font. This figure was created with *MarvinSketch* (ChemAxon).



**Figure 2**  
SDS–PAGE profile of the *A. nitroguajacolicus* Rü61a His<sub>6</sub>-HOD C69S/H251A protein. Lane 1, molecular-weight markers. The molecular weight of each protein band is shown in kDa on the left of the figure. Lane 2, purified His<sub>6</sub>-HOD C69S/H251A protein.

**Table 1**

Data-collection statistics.

Values in parentheses are for the highest resolution bin.

Data set	Crystal form A (Fig. 3 <i>b</i> )	Crystal form B (Fig. 3 <i>d</i> )
Beamline	ID14-EH1 (ESRF)	BM30 (ESRF)
Wavelength (Å)	0.935	1.3783
Detector	ADSC Q210 CCD	MAR CCD (165 mm)
Crystal-to-detector distance (mm)	265	180
Exposure time per degree of rotation (s)	25	80
Space group	Processed in <i>P422</i>	<i>P4</i> <sub>3,2,1</sub>
Unit-cell parameters (Å)	<i>a</i> = <i>b</i> = 165.77, <i>c</i> = 44.53	<i>a</i> = <i>b</i> = 153.79, <i>c</i> = 120.87
Resolution limits† (Å)	40.0–3.15 (3.26–3.15)	30.00–3.50 (3.63–3.50)
<i>I</i> / $\sigma$ ( <i>I</i> )	17.9 (4.8)	22.0 (4.4)
Observations	80435	300495
Independent reflections	11373 (1102)	17555 (1417)
Overall redundancy	7.07 (6.7)	17.1 (9.4)
<i>R</i> <sub>sym</sub> ‡ (%)	9.9 (41.8)	11.6 (37.2)
Completeness (%)	99.9 (100.0)	95.0 (78.5)
Molecules per ASU	2‡	3
Solvent content (%)	48.5	67.0

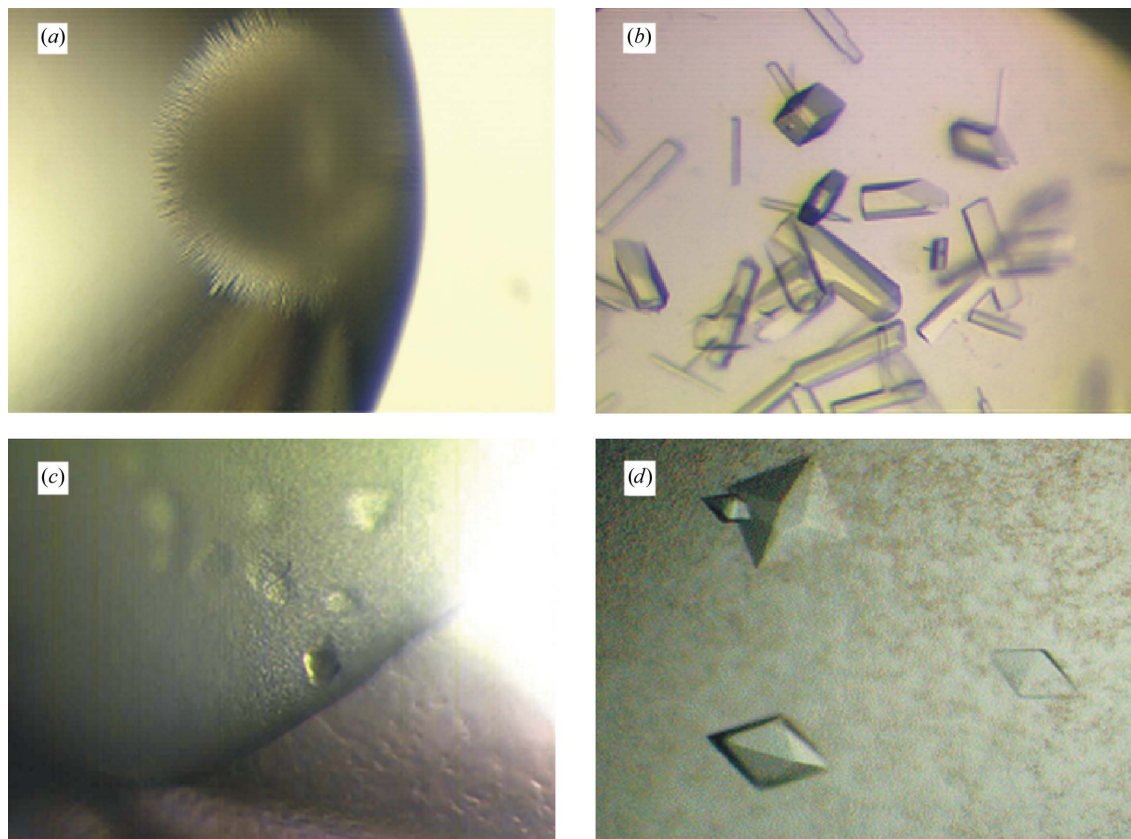
†  $R_{sym} = \sum_{hkl} \sum_i |I_{hkl,i} - \langle I(hkl) \rangle| / \sum_{hkl} \sum_i I_{hkl,i}$ , where  $\langle I(hkl) \rangle$  is the mean intensity of a set of equivalent reflections and  $I_{hkl,i}$  is the *i*th measurement of the reflection with Miller indices *hkl*. ‡ Given the twinned nature of crystal form A, predictions based on the Matthews coefficient analysis (Matthews, 1968) give four protein molecules per ASU, assuming the real point group to be either *P4* or *P222*.

malate, sodium malonate, sodium formate, ammonium nitrate, ammonium sulfate, sodium acetate, sodium chloride, lithium sulfate) at various concentrations yielded 'sea-urchin'-like microcrystals using 1.8 M sodium/potassium tartrate as a precipitant and His<sub>6</sub>-HOD C69S/H251A in storage buffer at a concentration of 50 mg ml<sup>-1</sup>

(Fig. 3*a*). Identical microcrystals were also obtained under the same conditions using the His<sub>6</sub>-HOD C69S variant.

Further crystal optimization was carried out using the His<sub>6</sub>-HOD C69S/H251A variant only. Single crystals were obtained by optimizing the precipitant and protein concentration as well as the pH used to obtain the microcrystals shown in Fig. 3*a*). The single crystals shown in Fig. 3*b*) (crystal form A) were obtained using His<sub>6</sub>-HOD C69S/H251A at a concentration of 150 mg ml<sup>-1</sup> in storage buffer in the presence of a reservoir composed of 1.65 M sodium/potassium tartrate and 0.1 M HEPES pH 7.0. SeMet-substituted His<sub>6</sub>-HOD C69S/H251A failed to crystallize under identical conditions. The His<sub>6</sub>-HOD C69S/H251A crystals obtained as described above diffracted to 3.15 Å resolution at the ID14-EH1 beamline (ESRF, Grenoble). Analysis of cumulative intensity distribution plots (data not shown) from diffraction data (Table 1, crystal form A) revealed these crystals, which appeared to belong to point group *P422*, exhibit twinning.

In parallel with attempts to solve the structure of His<sub>6</sub>-HOD C69S/H251A from twinned data using the molecular-replacement technique with various members of the  $\alpha/\beta$ -hydrolase-fold family as templates, we also tried to overcome the twinning problem by screening crystallization additives using the Additive Screening kit (Hampton Research). We found that the addition of 10 mM CuCl<sub>2</sub> to the 1.65 M sodium/potassium tartrate, 0.1 M HEPES pH 7.0 crystallization mixture yielded small His<sub>6</sub>-HOD C69S/H251A crystals of different morphology (Fig. 3*c*). Optimization of the additive concentration led to crystals of square bipyramidal shape (crystal form B; Fig. 3*d*) which typically grow to dimensions of 350 × 350 × 200 μm over three weeks. These His<sub>6</sub>-HOD C69S/H251A crystals grow from a protein solution at 150 mg ml<sup>-1</sup> in storage buffer in the



**Figure 3**  
Crystals of His<sub>6</sub>-HOD C69S/H251A obtained as described in §2.3.

presence of 1.65 M sodium/potassium tartrate, 0.1 M HEPES pH 7.0 and 30 mM CuCl<sub>2</sub>.

#### 2.4. X-ray diffraction

His<sub>6</sub>-HOD C69S/H251A crystals grown from 1.65 M sodium/potassium tartrate, 0.1 M HEPES pH 7.0 and 30 mM CuCl<sub>2</sub> were transferred for a few seconds into a reservoir containing 1.7 M sodium/potassium tartrate, 0.1 M HEPES pH 7.0, 30 mM CuCl<sub>2</sub> and 15% (v/v) glycerol and vitrified in liquid nitrogen for data collection. Although the high salt concentration present in the reservoir allows a high degree of cryoprotection, we observed improved behaviour in the presence of glycerol.

We reasoned that the crystallization of His<sub>6</sub>-HOD C69S/H251A in the presence of a high concentration of Cu<sup>2+</sup> ions might have resulted in heavy atoms bound to the histidine tag and that these might be useful for phasing purposes. A single anomalous dispersion (SAD) data set (Table 1, crystal form B) was collected at 3.5 Å resolution using synchrotron radiation at the Cu absorption edge (1.3783 Å). Crystallographic data for both HOD crystal forms were processed using the *HKL* suite (Otwinowski & Minor, 1997). Analysis of intensity statistics revealed that His<sub>6</sub>-HOD C69S/H251A crystals grown in the presence of CuCl<sub>2</sub> no longer exhibited signs of twinning.

### 3. Results

Data from HOD crystal form B were used for structure solution. In order to solve the structure of His<sub>6</sub>-HOD C69S/H251A, the graphical user interface *HKL2MAP* (Pape & Schneider, 2004) was used to run several programs (*SHELXC*, *SHELXD*, *SHELXE*) from the *SHELX* suite. After data preparation with *SHELXC* (Sheldrick, 2003), the substructure-solution program *SHELXD* (Schneider & Sheldrick, 2002) easily located six high-occupancy Cu atoms in 30 of the 100 phase trials performed. These sites were then directly passed to *SHELXE* (Sheldrick, 2002) for phasing and density modification (solvent content set to 65%). A clear difference in contrast was seen between the original and inverted-hand enantiomorphs. Inspection of the electron-density maps revealed regions of correct right-hand helices in space group *P*<sub>4</sub><sub>3</sub><sub>2</sub><sub>1</sub><sub>2</sub> which could be fitted to helical segments using the *ARP/wARP* HelixBuild module (Morris *et al.*, 2004). Model building using maps calculated from *SHARP* (Bricogne *et al.*, 2003) improved phases is nearly complete. We anticipate that

the Cu ions used for phasing bind to the N-terminal hexahistidine tags available and that HOD adopts the predicted  $\alpha/\beta$ -hydrolase-fold topology.

The staff scientists operating the beamlines ID14-EH1 and BM30 (European Molecular Biology Laboratory, Grenoble) are acknowledged for their help during the experiments. The York Structural Biology Laboratory (University of York, UK) and the Musacchio laboratory (IFOM-IEO Campus, Milan, Italy) are also thanked for their support.

### References

- Bauer, I., Max, N., Fetzner, S. & Lingens, F. (1996). *Eur. J. Biochem.* **240**, 576–583.
- Bricogne, G., Vornrhein, C., Flensburg, C., Schiltz, M. & Paciorek, W. (2003). *Acta Cryst.* **D59**, 2023–2030.
- Brzozowski, A. M. & Walton, J. (2001). *J. Appl. Cryst.* **34**, 97–101.
- Fetzner, S. (2002). *Appl. Microbiol. Biotechnol.* **60**, 243–257.
- Fischer, F., Künne, S. & Fetzner, S. (1999). *J. Bacteriol.* **181**, 5725–5733.
- Frerichs-Deeken, U., Ranguelova, K., Kappl, R., Huttermann, J. & Fetzner, S. (2004). *Biochemistry*, **43**, 14485–14499.
- Hamilton, G. A. (1974). *Molecular Mechanisms of Oxygen Activation*, edited by O. Hayaishi, pp. 405–451. New York: Academic Press.
- Hayaishi, O. (1974). Editor. *Molecular Mechanisms of Oxygen Activation*, pp. 1–28. New York: Academic Press.
- Massey, V. (1994). *J. Biol. Chem.* **269**, 22459–22462.
- Matthews, B. W. (1968). *J. Mol. Biol.* **33**, 491–497.
- Morris, R. J., Zwart, P. H., Cohen, S., Fernandez, F. J., Kakaris, M., Kirillova, O., Vornrhein, C., Perrakis, A. & Lamzin, V. S. (2004). *J. Synchrotron Rad.* **11**, 56–59.
- Nardini, M. & Dijkstra, B. W. (1999). *Curr. Opin. Struct. Biol.* **9**, 732–737.
- Otwinowski, Z. & Minor, W. (1997). *Methods Enzymol.* **276**, 307–326.
- Palfey, B. A., Ballou, D. P. & Massey, V. (1995). *Active Oxygen in Biochemistry*, edited by J. S. Valentine, C. S. Foote, A. Greenberg & J. F. Liebman, pp. 37–83. London: Blackie.
- Pape, T. & Schneider, T. R. (2004). *J. Appl. Cryst.* **37**, 843–844.
- Qi, R., Fetzner, S. & Oakley, A. J. (2007). *Acta Cryst.* **F63**, 378–381.
- Que, L. Jr (1999). *Bioinorganic Catalysis*, edited by J. Reedijk & E. Bouwman, pp. 269–321. New York: Marcel Dekker.
- Schneider, T. R. & Sheldrick, G. M. (2002). *Acta Cryst.* **D58**, 1772–1779.
- Sheldrick, G. M. (2002). *Z. Kristallogr.* **217**, 644–650.
- Sheldrick, G. M. (2003). *SHELXC*. University of Göttingen, Göttingen, Germany.
- Steiner, R. A., Kalk, K. H. & Dijkstra, B. W. (2002). *Proc. Natl Acad. Sci. USA*, **99**, 16625–16630.